Acta Crystallographica Section C Crystal Structure Communications ISSN 0108-2701

N-(*tert*-Butoxycarbonyl)- α -aminoisobutyryl- α -aminoisobutyric acid methyl ester: two polymorphic forms in the space group $P2_1/n$

Hadgu Girmay Gebreslasie,^{a,b} Øyvind Jacobsen^c and Carl Henrik Görbitz^a*

^aDepartment of Chemistry, University of Oslo, PO Box 1033 Blindern, N-0315 Oslo, Norway, ^bDepartment of Medicine-Medical Biochemistry, College of Health Sciences, Mekelle University, PO Box 1871, Mekelle-Tigray, Ethiopia, and ^cSchool of Pharmacy, University of Oslo, PO Box 1068 Blindern, N-0316 Oslo, Norway Correspondence e-mail: c.h.gorbitz@kjemi.uio.no

Received 6 June 2011 Accepted 21 June 2011 Online 5 July 2011

The title compound (systematic name: methyl 2-{2-[(tertbutoxycarbonyl)amino]-2-methylpropanamido}-2-methylpropanoate), C14H26N2O5, (I), crystallizes in the monoclinic space group $P2_1/n$ in two polymorphic forms, each with one molecule in the asymmetric unit. The molecular conformation is essentially the same in both polymorphs, with the α -aminoisobutyric acid (Aib) residues adopting φ and ψ values characteristic of α -helical and mixed 3_{10} - and α -helical conformations. The helical handedness of the C-terminal residue (Aib2) is opposite to that of the N-terminal residue (Aib1). In contrast to (I), the closely related peptide Boc-Aib-Aib-OBn (Boc is tert-butoxycarbonyl and Bn is benzyl) adopts an $\alpha_{\rm L}$ -P_{II} backbone conformation (or the mirror image conformation). Compound (I) forms hydrogen-bonded parallel β -sheet-like tapes, with the carbonyl groups of Aib1 and Aib2 acting as hydrogen-bond acceptors. This seems to represent an unusual packing for a protected dipeptide containing at least one α, α -disubstituted residue.

Comment

 α -Aminoisobutyric acid (Aib) is an achiral nonproteinogenic amino acid found in peptaibiotics, a group of fungal peptides with antibiotic activity (Degenkolb & Brückner, 2008; Toniolo & Brückner, 2009). Peptaibiotics, exemplified by alamethicin (Pandey *et al.*, 1977), are believed to exert their biological effect by folding into amphipathic helices that oligomerize, forming voltage-gated transmembrane ion channels (Mueller & Rudin, 1968; Nagaraj & Balaram, 1981; Fox & Richards, 1982). Key to the biological activity of peptaibiotics is the (conformational) preference of Aib for helical conformations. As was first recognized by Ramachandran & Chandrasekaran (1972) and, independently, by Marshall & Bosshard (1972), the Aib residue is almost invariably restricted to φ and ψ values corresponding to the right- ($\varphi = -60\pm 20^\circ$ and $\psi =$ $-30\pm20^\circ$) or left-handed ($\varphi = 60\pm20^\circ$ and $\psi = 30\pm20^\circ$) 3_{10} - or α -helical regions of the Ramachandran plot (Venkatraman et al., 2001). It has been known for a long time that Aib can increase the conformational stability of peptide helices (Burgess & Leach, 1973; Karle & Balaram, 1990) with both α and 310-helical hydrogen-bonding patterns (Marshall et al., 1990). The introduction of Aib residues into polypeptide chains limits the range of conformations accessible to the peptide because of the extra methyl group at the C^{α} atom, forcing the peptide chain into a left- or right-handed helical conformation or nucleating a β -turn (Aravinda *et al.*, 2003). Numerous X-ray diffraction studies of short Aib-based model peptides have demonstrated their preference for 310-helical structures (Karle & Balaram, 1990; Toniolo & Benedetti, 1991; Toniolo et al., 2001). A review of crystal structures of synthetic tri-, tetra- and pentapeptides containing at least one Aib residue showed that almost all form incipient 310-helices (Toniolo et al., 1983). However, while shorter Aib peptides preferentially adopt type III/III' β -turn and 3₁₀-helical conformations, longer Aib peptides are able to form α -helical structures (Butters et al., 1981; Schmitt et al., 1982; Pavone et al., 1990).

Although 3_{10} - and α -helical conformations are statistically by far the most prevalent conformations observed for Aib in crystal structures of Aib-containing peptides, a number of Aib residues have also been found to adopt polyproline II conformations, in particular in structures of protected di- and tripeptides (Aravinda *et al.*, 2008). Other nonhelical conformations are, however, very rare. Notably, because of the severe steric clash between the carbonyl group of the preceding residue and one of the methyl groups, β -strand conformations are energetically very unfavourable (Aravinda *et al.*, 2008), making Aib one of the best β -sheet-breaking amino acids (Moretto *et al.*, 1989; Toniolo *et al.*, 2001).

Aib residues at the *C*-terminus of a helix have a tendency to adopt a different conformation from the rest of the molecule. In a recent investigation of 143 crystal structures of Aib-containing helical peptides with more than three residues, 66.2% of the *C*-terminal Aib residues were found to adopt helical conformations corresponding to a different helical handedness than the body of the peptide, and 20.3% to adopt polyproline II conformations (Aravinda *et al.*, 2008).

The title compound, (I), was synthesized as part of an ongoing effort to develop a generic methodology for the conformational stabilization of synthetic analogues of 3_{10} -helical protein segments (Jacobsen *et al.*, 2009, 2011).



Many biologically important protein–protein interactions are mediated by helical protein segments and could therefore, in principle, be modulated by synthetic peptides with similar



Figure 1

The molecular structure of (I) in polymorph A at 105 K (top) and in polymorph B at 293 K (bottom). The atomic numbering scheme is the same for both polymorphs. Displacement ellipsoids are drawn at the 50% probability level.

primary structures and conformations. Because the entropy reduction associated with ligand-receptor binding is likely to be smaller for a prestructured or conformationally restricted peptide than for a related random coil peptide, the ability of Aib to induce or stabilize helical conformations makes Aibcontaining peptides mimicking protein segments potentially valuable in drug discovery. Improved proteolytic stability is another beneficial effect of a higher degree of helicity in solution (Banerjee *et al.*, 2002), which derives from the fact that proteases recognize their substrates in a β -strand conformation (Tyndall *et al.*, 2005).

Diffraction data were collected for two needle-shaped crystals, which proved to represent two different polymorphs of (I), hereafter denoted A and B, which both belong to the monoclinic space group $P2_1/n$ (see *Experimental*). The molecular structure of (I) is depicted in Fig. 1. The conformation is virtually the same in both polymorphic forms, as reflected by the torsion angles listed in Table 1 and the r.m.s. deviation of 0.157 Å for the best fit between heavy atoms. The φ and ψ values are characteristic of α -helical and mixed 3₁₀- and α helical conformations. An α -helix is defined by the presence of two or more consecutive $i \rightarrow i + 4$ intramolecular hydrogen bonds and thus involves at least six residues. Similarly, two or more consecutive $i \rightarrow i+3$ intramolecular hydrogen bonds constitute the defining feature of a 3_{10} -helix. It is difficult to label the conformation of a single isolated residue as 3_{10} - or α -helical exclusively based on its torsion angles. For the purpose of this study, we define the backbone conformation of



Figure 2

(a) The hydrogen-bonded tape parallel to the shortest crystallographic axis (about 6.1 Å) occurring in both polymorphs of (I) (the drawing is for form A). (b) The hydrogen-bonded tape in the structure of Boc-Aib-L-Ile-OMe (CSD refcode AJOLEQ; Nilofarnissa *et al.*, 2000).

a single isolated residue to be α -helical if a hypothetical oligopeptide with the same torsion angles as the said residue would be α -helical, *i.e.* would form consecutive $i \rightarrow i + 4$ intramolecular hydrogen bonds. Notably, the helical handedness of the C-terminal residue Aib2 is opposite to that of the N-terminal residue Aib1 in both polymorphs, which helps to avoid unfavourable intramolecular contacts (Van Roey et al., 1983) [as (I) is achiral and crystallizes in a centrosymmetric space group, it is not meaningful to designate the conformations of Aib1 and Aib2 as left- or right-handed]. Interestingly, Aib2 in the closely related protected dipeptide Boc-Aib-Aib-OBn [Cambridge Structural Database (CSD; Version 5.32 of November 2010; Allen, 2002) refcode BAJROT10 (Van Roey et al., 1983); Table 1] adopts a polyproline II conformation (or its mirror image conformation) instead of a 3_{10} - or α -helical conformation.

Pairs of strong hydrogen bonds (Tables 2 and 3) link the peptide molecules of (I) into tapes, as shown in Fig. 2(*a*). In form *A*, these are supported by two reasonably linear C– $H \cdot \cdot \cdot O$ =C contacts with $H \cdot \cdot \cdot O < 2.60$ Å; these are essentially missing in form *B* as the pertinent $H \cdot \cdot \cdot O$ distances are >2.85 Å. A similar tape motif has previously only been found for Boc- α -methyl-L-Phe-L-Val-OBn (CSD refcode CAPZIC; Van Roey *et al.*, 1981). Other protected Aib*-Xaa dipeptides in the CSD, where Aib* is either Aib or another α, α -disubstituted amino acid and Xaa is a chiral amino acid, form a second type of tape motif that in a sense constitutes a 'frame shift' compared with (I), as the pair of carbonyl acceptors is



Figure 3

The crystal packing of (I) in (a) polymorph A and (b) polymorph B, both viewed along the short a axis.

shifted one residue towards the *N*-terminal end of the peptide, corresponding to atoms O2 and O3 in Fig. 1 rather than atoms O3 and O4 used for (I) (Tables 2 and 3). Furthermore, the Aib* residue in every second molecule in the tape adopts a conformation corresponding to the opposite helical handedness (Fig. 2b and Table 1). All such structures have two molecules in the asymmetric unit. The same hydrogen-bonding pattern is also observed for the benzyl ester analogue of (I) (CSD refcode BAJROT10; Van Roey *et al.*, 1983) and for one out of three Xaa-Aib* peptides. Molecules in Table 1 with two α,α -disubstituted amino acid residues, where at least one is different from Aib, are evidently too crowded to form tape motifs and instead form various simple hydrogen-bonded chains.

Recent studies have revealed the importance of $n \rightarrow \pi^*$ C=O_i \rightarrow C_{i+1}=O hyperconjugative interactions between consecutive amide groups in stabilizing 3₁₀-helical, α -helical and polyproline II conformations (Bretscher *et al.*, 2001; Hodges & Raines, 2006; Jakobsche *et al.*, 2010), in particular when the distance *d* from atom O_i to atom C_{i+1} is less than 3.2 Å (Bartlett *et al.*, 2010). In a 3₁₀-helix, one such interaction can be worth as much as 1.3 kcal mol⁻¹ (5.4 kJ mol⁻¹; 1 kcal mol⁻¹ = 4.184 kJ mol⁻¹) (Bartlett *et al.*, 2010). If an ester is the electron-density acceptor, it has been found that an $n \rightarrow \pi^*$ C=O_i \rightarrow C_{i+1}=O interaction can provide 0.7 kcal mol⁻¹ of stabilization energy (Hinderaker & Raines, 2003). The observed O_i to C_{i+1} distances are 2.916 (2) and 2.663 (2) Å for Aib1 and Aib2, respectively, in polymorph A, while the corresponding values for polymorph B are 2.861 (3) and 2.730 (3) Å. The crystal structures of (I) thus provide good examples of $n \rightarrow \pi^*$ C=O_i \rightarrow C_{i+1}=O stabilizing interactions in a short peptide. A helical or polyproline II conformation allows the lone pair on the carbonyl O atom to interact with the antibonding C_{i+1} — O orbital along an angle of attack very close to the Bürgi–Dunitz angle (107°), the preferred angle of attack of a nucleophile at a carbonyl group (Bürgi et al., 1973). Significantly, the $O_i - C = O_{i+1}$ angles for both polymorphic forms of (I) are very close to the Bürgi-Dunitz angle, with values of 108.80 (11) and 103.01 (11)° for Aib1 and Aib2, respectively, in polymorph A, and 107.42 (16) and 104.08 (18)°, respectively, in polymorph B.

The overall crystal packing arrangements of the two polymorphs, illustrated in Fig. 3, are quite different, despite the occurrence of hydrogen-bonded tapes in both forms, as shown in Fig. 2. The B form is more clearly divided into layers.

Experimental

 α -Aminoisobutvric acid methyl ester hydrochloride was obtained by treating aminoisobutvric acid with thionvl chloride in methanol solution (Jacobsen et al., 2011). Compound (I) was synthesized by standard solution-phase peptide coupling of α -aminoisobutyric acid methyl ester, which was generated in situ from α -aminoisobutyric acid methyl ester hydrochloride by treatment with N,N-diisopropylethylamine, with commercially available N-(tert-butoxycarbonyl)- α -aminoisobutyric acid. 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDC) was used as coupling reagent and 1.0 equivalents of 1-hydroxybenzotriazole (HOBt) was added to catalyse the reaction (Jacobsen et al., 2011). A small quantity of (I) (about 5 mg) was dissolved in ethyl acetate (30 µl). Needle-shaped crystals appeared as water vapour diffused into the solution at room temperature. Data were first collected under ambient conditions because of a temporary failure of the low-temperature device. When the cooling unit was available again, data were recorded for a second crystal taken from the same batch. Although there were no obvious differences in appearance, this crystal proved to be a different polymorph. We thus had data for two concomitant forms, A (data collected at low temperature) and B(ambient). Several other crystals were subsequently tested to find a good specimen for collection of a low-temperature data set for form B (the original crystal had unfortunately been lost), but only crystals of form A were found, suggesting that A was the dominant polymorph in the crystalline sample. Crystals of form A can also be cooled and heated without being converted to form B.

Polymorph A of (I)

Crystal data	
$C_{14}H_{26}N_2O_5$	$V = 1704.7 (13) \text{ Å}^3$
$M_r = 302.37$	Z = 4
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 6.116 (3) \text{ Å}_{a}$	$\mu = 0.09 \text{ mm}^{-1}$
b = 15.662 (7) Å	$T = 105 { m K}$
c = 17.967 (8) Å	$0.78 \times 0.22 \times 0.13~\mathrm{mm}$
$\beta = 97.878 \ (6)^{\circ}$	

Table 1

Main torsion angles (°) in the crystal structures of protected dipeptides in the CSD.

Abbreviations: Boc is tert-butoxycarbonyl, Cbz is carboxybenzyl, B = Aib, B* is another α, α -disubstituted amino acid, I is isoleucine, V is valine, A is alanine, F is phenylalanine, Me is methyl, Bn is benzyl, 'B is tert-butyl and L is leucine.

Compound or CSD refcode	Sequence	$arphi_1$ †	ψ_1	$arphi_2$	ψ_2	Inverse‡
Polymorph (IA)	Boc-B-B-OMe	58.96 (19)	40.82 (19)	-45.6(2)	-50.47 (18)	
Polymorph (IB)	Boc-B-B-OMe	56.4 (3)	42.5 (3)	-50.5(3)	-49.1 (3)	
BAJROT10	Boc-B-B-OBn	59.6	52.0	-51.5	138.4	i
PARDUH	Boc-B-B*-OMe	61.5	32.8	177.0	-179.4	
		62.6	33.3	177.3	179.2	
PUXHOF	Boc-B*-B*-OMe	55.0	42.7	-56.0	-34.7	i
VEYQAR	Cbz-B*-B*-OMe	65.5	26.6	-35.5	-50.8	i
AJOLEQ	Boc-B-I-OMe	63.9	45.8	-68.0	-29.8	
		57.2	44.3	122.1	-148.1	i
CAPZIC	Boc-B*-V-OBn	58.9	33.3	-57.2	-44.2	
GANPEQ	Cbz-B-A-O'B	57.8	41.8	-78.9	170.54	
		58.8	46.3	133.3	-174.9	i
OBAZIA	Boc-B*-A-OMe	56.1	45.8	-95.1	-177.9	i
		58.4	42.8	145.8	-24.4	
OFUXOC	Boc-B*-L-OMe	60.8	47.1	-72.3	-36.2	
		64.3	48.7	122.7	-9.6	i
PASGUL	Boc-B-F-OMe	62.1	44.1	-75.9	1.6	
		53.7	50.7	122.0	-164.0	i
XOWVAG	Boc-B-L-OMe	58.1	46.7	-80.7	-0.5	
		58.4	44.5	112.9	-165.7	i
LAGFII	Boc-V-B*-OMe	73.2	-127.3	55.2	32.2	i
TIRJOT	Cbz-A-B-O'B	76.6	-156.2	-51.0	-44.7	i
ZAQVUI	Cbz-L-B*-OMe	90.6	32.2	-43.7	-53.1	i

 \dagger For (I), with reference to Fig. 1, the listed torsion angles are: $\varphi_1 = C5 - N1 - C6 - C9$, $\psi_1 = N1 - C6 - C9 - N2$, $\varphi_2 = C9 - N2 - C10 - C13$ and $\psi_2 = N2 - C10 - C13 - O5$. \ddagger To facilitate comparison between structures, molecules indicated by 'i' have been inverted to obtain the same sign for the first torsion angle φ_1 .

Data collection

Bruker APEXII CCD area-detector diffractometer Absorption correction: multi-scan (SADABS; Bruker, 2007) $T_{\rm min} = 0.917, \ T_{\rm max} = 0.988$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.050$ H atoms treated by a mixture of $wR(F^2) = 0.126$ S = 1.02refinement $\Delta \rho_{\text{max}} = 0.28 \text{ e} \text{ Å}^{-3}$ 4100 reflections $\Delta \rho_{\rm min} = -0.24 \text{ e} \text{ Å}^{-3}$ 196 parameters

Polymorph B of (I)

Crystal data

C14H26N2O5 $M_{\rm m} = 302.37$ Monoclinic, $P2_1/n$ a = 6.0679 (15) Åb = 33.743 (9) Åc = 8.583 (2) Å $\beta = 92.266 (3)^{\circ}$

Data collection

Bruker APEXII CCD area-detector diffractometer Absorption correction: multi-scan (SADABS; Bruker, 2007) $T_{\min} = 0.814, T_{\max} = 0.991$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.042$ wR(F²) = 0.125 S = 1.003105 reflections 197 parameters

11873 measured reflections 4100 independent reflections 2595 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.048$

independent and constrained

V = 1756.0 (8) Å ³
Z = 4
Mo $K\alpha$ radiation
$\mu = 0.09 \text{ mm}^{-1}$
T = 293 K
$0.80 \times 0.11 \times 0.10 \text{ mm}$

10165 measured reflections 3105 independent reflections 1584 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.052$

H atoms treated by a mixture of
independent and constrained
refinement
$\Delta \rho_{\rm max} = 0.15 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.18 \text{ e} \text{ Å}^{-3}$

Table 2

Hydrogen-bond geometry (Å, $^{\circ}$) for polymorph A of (I).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N1 - H1 \cdots O3^{i}$ $N2 - H2 \cdots O4^{i}$ $C3 - H32 \cdots O2^{i}$ $C11 - U112 - O1^{ii}$	0.840 (19)	2.12 (2)	2.963 (2)	175.6 (18)
	0.86 (2)	2.44 (2)	3.227 (2)	151.6 (17)
	0.98	2.51	3.298 (2)	137

Symmetry codes: (i) x - 1, y, z; (ii) x + 1, y, z.

Table 3

Hydrogen-bond geometry (Å, $^{\circ}$) for polymorph *B* of (I).

$D - H \cdots A$	$D-{\rm H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$\begin{array}{c} N1 {-} H1 {\cdots} O3^i \\ N2 {-} H2 {\cdots} O4^i \end{array}$	0.86 (2)	2.16 (2)	3.008 (3)	170 (2)
	0.85 (2)	2.44 (2)	3.197 (3)	148 (2)

Symmetry code: (i) x + 1, y, z.

Positional parameters were refined for H atoms bonded to N atoms. Methyl H atoms were positioned with idealized geometry and fixed at C-H = 0.98 (form A, 105 K) or 0.96 Å (form B, 293 K). $U_{\rm iso}({\rm H})$ values were set at $1.2U_{\rm eq}({\rm N})$ for N-H groups or at $1.5U_{\rm eq}({\rm C})$ for methyl groups.

For both polymorphs, data collection: APEX2 (Bruker, 2007); cell refinement: SAINT-Plus (Bruker, 2007); data reduction: SAINT-Plus. Program(s) used to solve structure: SHELXS97 (Sheldrick, 2008) for polymorph A; SHELXTL (Sheldrick, 2008) for polymorph B. Program(s) used to refine structure: SHELXL97 (Sheldrick, 2008) for polymorph A; SHELXTL for polymorph B. Molecular graphics: SHELXL97 for polymorph A; SHELXTL for polymorph B. Software used to prepare material for publication: SHELXL97 for polymorph A; SHELXTL for polymorph B.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: JZ3206). Services for accessing these data are described at the back of the journal.

References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Aravinda, S., Shamala, N. & Balaram, P. (2008). Chem. Biodivers. 5, 1238–1262.
- Aravinda, S., Shamala, N., Roy, R. S. & Balaram, P. (2003). Proc. Indian Acad. Sci. Chem. Sci. 115, 373–400.
- Banerjee, R., Basu, G., Chene, P. & Roy, S. (2002). J. Pept. Res. 60, 88-94.
- Bartlett, G. J., Choudhary, A., Raines, R. T. & Woolfson, D. N. (2010). *Nat. Chem. Biol.* 6, 615–620.
- Bretscher, L. E., Jenkins, C. L., Taylor, K. M., DeRider, M. L. & Raines, R. T. (2001). J. Am. Chem. Soc. 123, 777–778.
- Bruker (2007). APEX2, SAINT-Plus and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
- Burgess, A. W. & Leach, S. J. (1973). Biopolymers, 12, 2599-2605.
- Bürgi, H. B., Dunitz, J. D. & Shefter, E. (1973). J. Am. Chem. Soc. 95, 5065-5067.
- Butters, T., Hütter, P., Jung, G., Pauls, N., Schmitt, H., Sheldrick, G. M. & Winter, W. (1981). Angew. Chem. Int. Ed. 20, 889–890.
- Degenkolb, T. & Brückner, H. (2008). Chem. Biodivers. 5, 1817–1843.
- Fox, R. O. & Richards, F. M. (1982). Nature (London), 300, 325-330.
- Hinderaker, M. P. & Raines, R. T. (2003). Protein Sci. 12, 1188-1194.
- Hodges, J. A. & Raines, R. T. (2006). Org. Lett. 8, 4695-4697
- Jacobsen, Ø., Klaveness, J., Ottersen, O. P., Amiry-Moghaddam, M. R. & Rongved, P. (2009). Org. Biomol. Chem. 7, 1599–1611.
- Jacobsen, Ø., Maekawa, H., Ge, N.-H., Görbitz, C. H., Rongved, P., Ottersen, O. P., Amiry-Moghaddam, M. & Klaveness, J. (2011). J. Org. Chem. 76, 1228–1238.
- Jakobsche, C. E., Choudhary, A., Miller, S. J. & Raines, R. T. (2010). J. Am. Chem. Soc. 132, 6651–6653.
- Karle, I. L. & Balaram, P. (1990). Biochemistry, 29, 6747-6756.

- Marshall, G. R. & Bosshard, H. E. (1972). Circ. Res. **30–31**, Suppl. II, 143–150.
- Marshall, G. R., Hodgkin, E. E., Langs, D. A., Smith, G. D., Zabrocki, J. & Leplawy, M. T. (1990). Proc. Natl Acad. Sci. USA, 87, 487–491.
- Moretto, V., Crisma, M., Bonora, G. M., Toniolo, C., Balaram, H. & Balaram, P. (1989). Macromolecules, 22, 2939–2944.
- Mueller, P. & Rudin, D. O. (1968). Nature (London), 217, 713-719.
- Nagaraj, R. & Balaram, P. (1981). Acc. Chem. Res. 14, 356-362.
- Nilofarnissa, M., Banumathi, S., Velmurugan, D. & Ramasubbu, N. (2000). Cryst. Res. Technol. 35, 333-341.
- Pandey, R. C., Cook, J. C. Jr & Rinehart, K. L. Jr (1977). J. Am. Chem. Soc. 99, 8469–8483.
- Pavone, V., Benedetti, E., Di Blasio, B., Pedone, C., Santini, A., Bavoso, A., Toniolo, C., Crisma, M. & Sartore, L. (1990). J. Biomol. Struct. Dyn. 7, 1321– 1331.
- Ramachandran, G. N. & Chandrasekaran, R. (1972). Progress in Peptide Research, Vol. II, Proceedings of the Second American Peptide Symposium, Cleveland, 1970, edited by S. Lande, p. 195. New York: Gordon & Breach.
- Schmitt, H., Winter, W., Bosch, R. & Jung, G. (1982). *Liebigs Ann. Chem.* pp. 1304–1321.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Toniolo, C. & Benedetti, E. (1991). Trends Biochem. Sci. 16, 350-353.
- Toniolo, C., Bonora, G. M., Bavoso, A., Benedetti, E., Di Blasio, B., Pavone, V. & Pedone, C. (1983). *Biopolymers*, 22, 205–215.
- Toniolo, C. & Brückner, H. (2009). In *Peptaibiotics: Fungal Peptides* Containing α-Dialkyl α-Amino Acids. New York: Wiley-VCH.
- Toniolo, C., Crisma, M., Formaggio, F. & Peggion, C. (2001). *Biopolymers*, 60, 396–419.
- Tyndall, J. D. A., Nall, T. & Fairlie, D. P. (2005). Chem. Rev. 105, 973-1000.
- Van Roey, P., Smith, G. D., Balasubramanian, T. M. & Marshall, G. R. (1981).
- Acta Cryst. B**37**, 1785–1788. Van Roey, P., Smith, G. D., Balasubramanian, T. M. & Marshall, G. R. (1983). Acta Cryst. C**39**, 894–896.
- Venkatraman, J., Shankaramma, S. C. & Balaram, P. (2001). Chem. Rev. 101, 3131–3152.